

Dispersions Containing Pyrogenic Oxides

Cross Reference to Related Applications

The present application claims priority to European application 00107817.9 filed on April 12, 2000, the subject matter of which is hereby incorporated by reference.

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Field of the Invention

The present invention is directed to dispersions, a process for preparing these dispersions and their use in the preparation of coating mixtures for inkjet media.

10 Background of the Invention

It is known that dispersions, for example water-based dispersions, can be prepared from pyrogenically prepared oxides. Aqueous dispersions can be used to prepare coating mixtures which are applied to paper or films. The coated films may then be printed using an inkjet printer. In this case, one objective is to obtain dispersions which are filled as highly as possible (high solids content) but with a low viscosity.

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Summary of the Invention

The invention provides dispersions which are characterised in that they consist of a liquid phase, preferably water, and a solid phase. The solid phase consists of a pyrogenic oxide, doped using an aerosol, the BET surface area of which is between 5 and 600 m²/g. The pyrogenic oxide preferably comprises silica prepared by the method of flame hydrolysis or flame oxidation and which has been doped with one or more doping component(s). Preferably the doping component is an aluminium oxide, and doping is accomplished by the method of aerosol application, in which the amount of doped material is between 1 and 200,000 ppm and the doping component(s) are applied via a salt or a salt mixture. The solid phase in the dispersion should be present in a proportion by weight between 0.001 and 80 wt.%.
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The invention also provides a process for preparing chemical dispersions in which a pyrogenic oxide, doped using an aerosol, is introduced into a liquid, preferably water, in a proportion by weight of between 0.001 and 80 wt.%. This dispersion is then
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that patent. 85 kg/h of SiCl_4 are evaporated, mixed with 51 Nm^3/h of hydrogen and with 70 Nm^3/h of a nitrogen/oxygen mixture (containing 35 vol.% O_2 , remainder N_2) and fed into the central tube in the burner. The gas mixture flows out of a nozzle and burns in a water-cooled combustion chamber. 4 Nm^3/h of (jacket) hydrogen flows out of the jacket nozzle which surrounds the central nozzle, in order to avoid incrustations. 70 Nm^3/h of secondary air are also introduced into the combustion chamber.

An aerosol flows into the central tube out of an axial tube located therein. The aerosol is produced by atomising a 15% aqueous AlCl_3 solution using a two-fluid nozzle. An aerosol flow of 1 kg/h (aqueous salt solution) is produced in which a carrier gas stream of 16 Nm^3/h of air conveys the aerosol through a heating section. The air/aerosol gas mixture then enters the central tube from the axial tube at about 180 °C. The aerosol is burned together with the air/ SiCl_4 mixture. The reaction gases and the pyrogenically prepared silica doped with aluminium oxide are removed under suction through a cooling system, and cooled by applying a reduced pressure. The solid material (doped pyrogenic oxide) is separated from the gas stream in a filter or a cyclone.

The doped pyrogenically prepared silica is produced as a white, finely divided powder. Adhering residues of hydrochloric acid are removed by treatment at elevated temperature with air which contains water vapour. The pyrogenic silicon dioxide doped using an aerosol has the following physico-chemical characteristics:

BET:	60 m^2/g
pH (4% aqu. disp.):	3.9
Compacted bulk density:	142 g/l
Chloride content:	180 ppm
Al_2O_3 content	0.19 wt. %
DBP absorption:	73 g/100 g
(DBP: dibutyl phthalate)	

An aqueous dispersion is prepared with the doped pyrogenic oxide. A commercially available Aerosils (pyrogenically prepared silica) provided by Degussa-Hüls AG /Frankfurt, OX 50 and Aerosil 90, are used as comparison examples.

Table 1 gives the characteristics of the oxides:

Table 1

Physico-chemical characteristics of the doped pyrogenic oxide

	Doped pyrogenic oxide according to example	OX 50	Aerosil 90
BET m ² /g	60	50	90
pH (4% aqu. disp.)	3.9	3.8 – 4.8	3.7 – 4.7
Compacted bulk density g/l	142	130	80
Chloride content ppm	180	< 250	< 250
Al ₂ O ₃ content wt. %	0.19	< 0.08	0.05
SiO ₂ content wt. %	99.8	> 99.8	> 99.8

5 An aqueous dispersion is prepared using these three different pyrogenic oxides. This is achieved using a rotor-stator system (Ultra-Turrax™) with a dispersion time of 30 minutes in a double-walled vessel with water cooling. It is attempted to prepare a 40% (with respect to solids) dispersion (w= 0.40). This dispersion may also be prepared using other equipment, *e.g.*, ball mills or pearl mills or various types of jet or high-
10 pressure mills (jets of liquid directed towards each other). It is shown that it is not possible to produce a 40% dispersion with Aerosil 90 using this system because the system is too highly viscous. The viscosity of the dispersions prepared in this way (doped oxide and Aerosil OX 50) is measured after 2 h with a Brookfield viscometer.

Table: 2

Viscosity of the 40% aqueous dispersion

	Doped pyrogenic oxide according to example 1	OX 50	Aerosil 90: not possible to prepare a 40% dispersion using Ultra-Turrax
5 rpm	2420	2320	> 10,000
10 rpm	1520	1320	
20 rpm	970	745	
50 rpm	554	372	
100 rpm	370	256	

rpm = revolutions per minute of the Brookfield viscometer

Inkjet coating mixtures are prepared from these 40% aqueous dispersions. Formulation for preparing an inkjet coating mixture is as follows. Two dispersions, A and B, are prepared. Dispersion A is a 40% ($w = 0.40$) aqueous dispersion which contains the pyrogenic oxide (or the doped pyrogenic oxide). This is made by dispersing the pyrogenic oxide or doped oxide for 30 minutes with an Ultra-Turrax system in a water-cooled double-jacket system.

Dispersion B is a 10% (with respect to PVA) aqueous dispersion of polyvinyl alcohol (solid, abbreviated as PVA), Mowiol 26-88 from the Clariant Co. The two dispersions A and B are combined over the course of 10 minutes by stirring at 500 rpm with a dissolver disc to give a dispersion C. Dispersions A and B are mixed in such a way that a ratio by weight of Aerosil (or doped pyrogenic oxide) to PVA of 100:20 is produced in subsequent dispersion C. In the case of a 40% dispersion A, this is mixed with dispersion B in the ratio by weight of 1.25:1 in order to achieve the required ratio by weight (100:20 for the solids). Furthermore (if required) enough water is added to produce a 24% dispersion C, with respect to the sum of the solids (pyrogenic oxide + PVA). The viscosity of this dispersion C, the inkjet coating mixture, is measured after 24 h using a Brookfield viscometer.

Table 3

Viscosity of the coating mixture measured after 24 h

	Doped oxide according to example 1	OX 50	Aerosil 90
Solids content of the coating mixture (pyrogenic oxide + PVA) wt. %	24	24	22.5
Viscosity [mPas] at 100 rpm	3244	685	3352

(Note: In the case of preparing the coating mixture from Aerosil 90, a 30% aqueous dispersion is used initially.)

These coating mixtures are applied to an untreated polyester film (thickness 100 micrometres) with the aid of a shaped spreading rod. The wet film thickness of the coating mixture is 120 micrometres. The coating is dried at 105°C for 8 minutes. The

also very good. Although dispersions made with Aerosil OX 50 also have a relatively low viscosity, the print quality of the coating mixtures, or coatings, prepared therefrom is not acceptable.

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